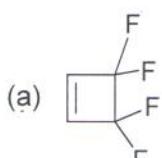


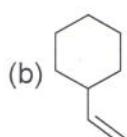
NAME KEY

1. Classify the following alkenes as electron rich or electron poor, relative to ethene

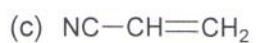
5



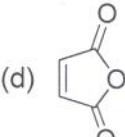
poor



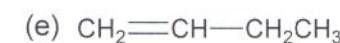
rich



poor



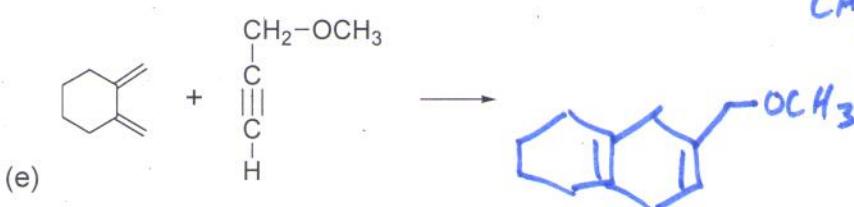
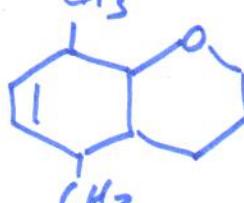
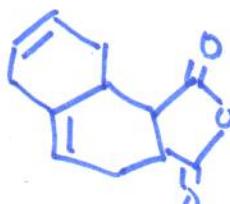
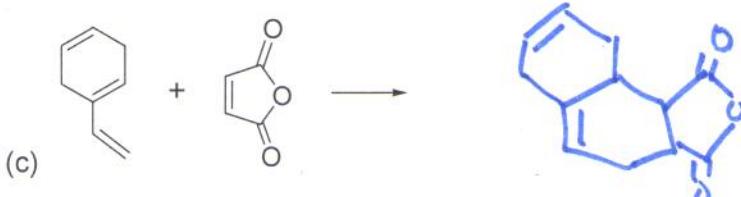
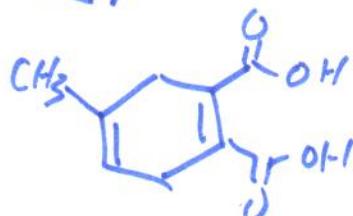
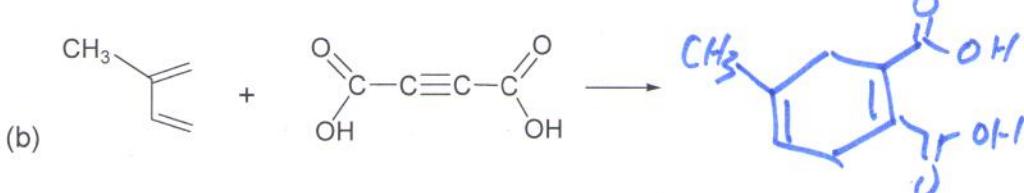
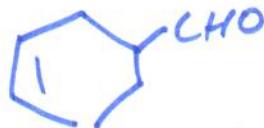
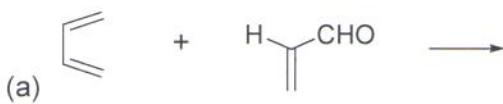
poor



rich

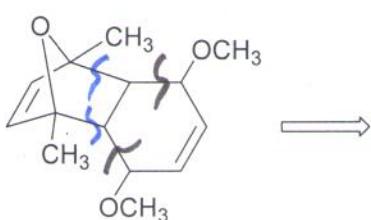
5

2. Draw the products from these Diels-Alder reactions:

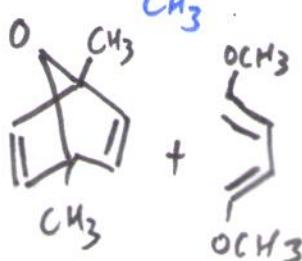
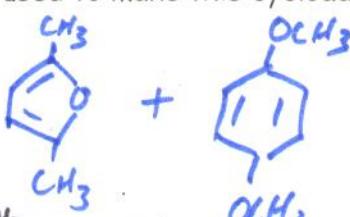


2

3. Draw the structure of the diene and dienophile used to make this cycloadduct:



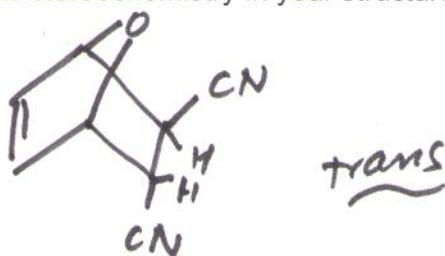
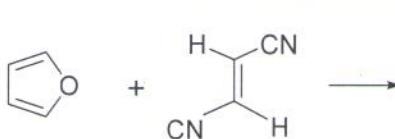
? + ?



12

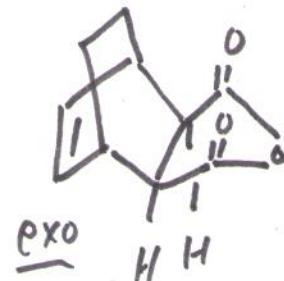
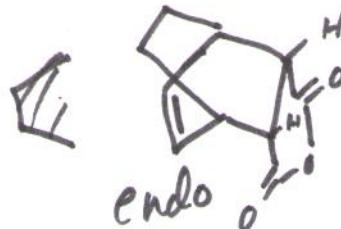
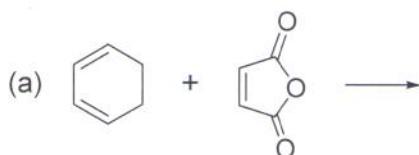
4. Draw the Diels-Alder cycloadduct and indicate stereochemistry in your structure.

2

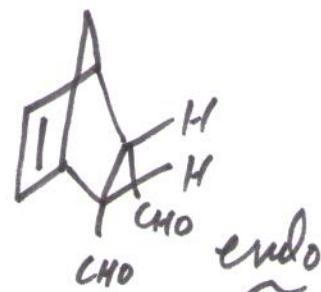
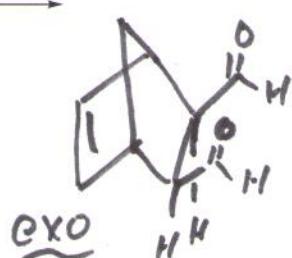
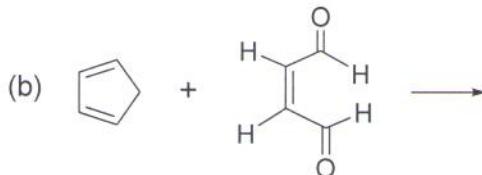


5. Draw both product isomers and indicate which is the endo and which is the exo cycloadduct.

2

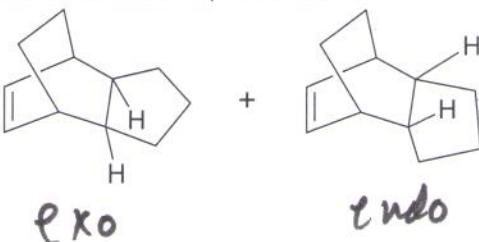
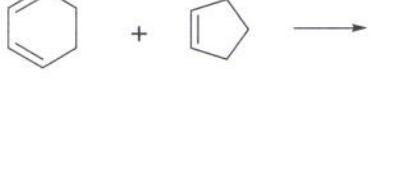


2



6. Indicate which is the endo and which is the exo cycloadduct.

1

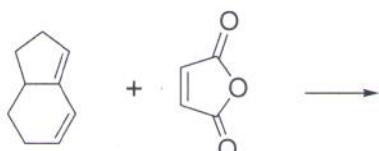


exo

endo

6. Explain why this Diels-Alder reaction will not work:

1



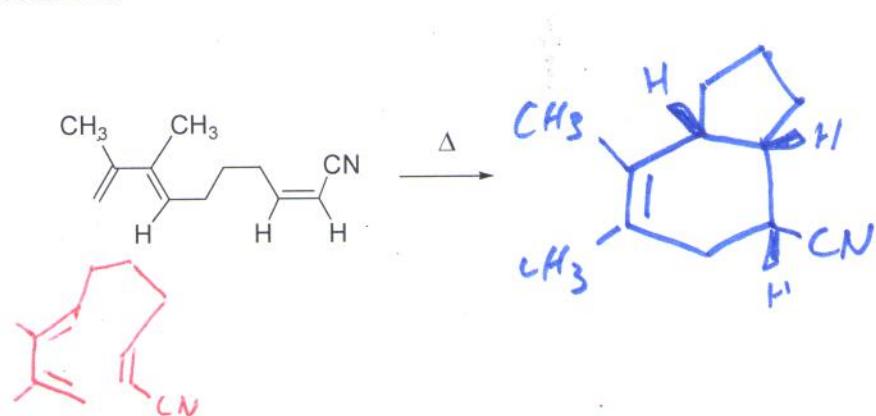
diene must be s-cis conformation

18

bicyclic ring system prevents this

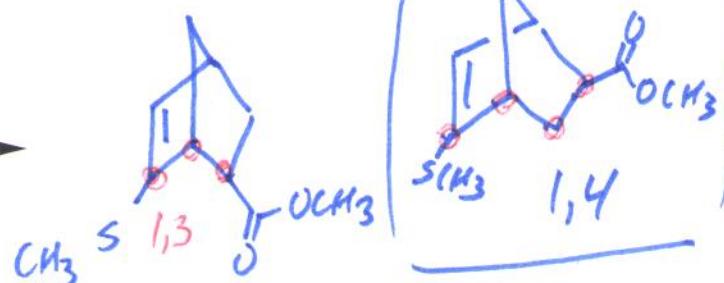
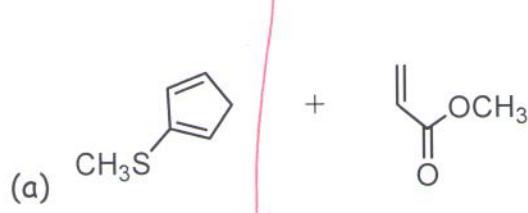
7. Draw the intramolecular DA cycloadduct. You may benefit from using a model for these cyclizations.

2



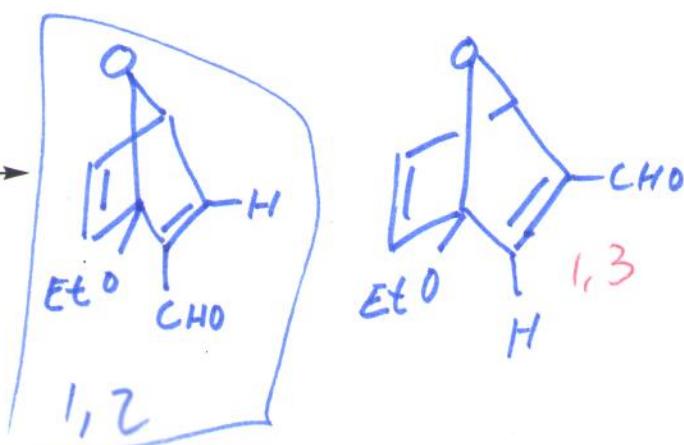
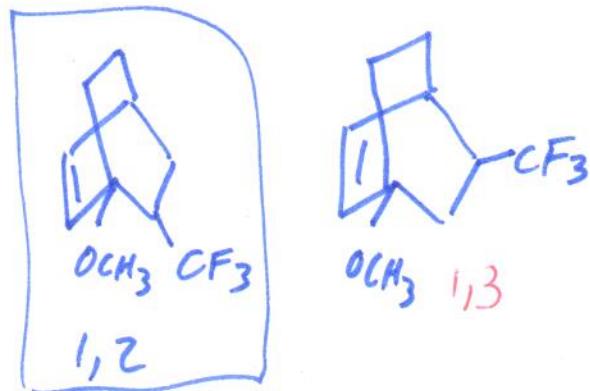
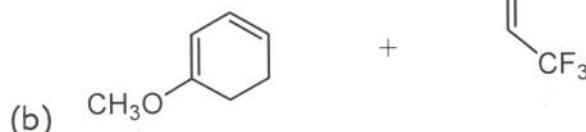
3

8. Predict which regio isomers will form using these unsymmetrical reagents:



1,4 vs 1,3
1,2 vs 1,3

[1,3 not preferred]



5